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## PECULIARITIES OF THE MINERAL COMPOSITION AND ORIGIN OF THE SIILLINJARVI CARBONATITES (FINLAND)

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The article presents the results of investigations of the Precambrian carbonatites (alvikites and beforecites) from the Siillinjarvi deposit (Finland). The carbonatites intruded into the older phlogopite-bearing rocks – syenites and glimmerites. Dolomite and calcite are the main minerals of the carbonatites (more than 70 % of the total carbonatite volume). Apatite, tetraferriphlogopite and richterite are the secondary minerals. The accessory minerals are presented by strontianite, monazite, pyrochlore, baddeleyite, ilmenite, magnetite and pyrite. We established the stable chemical composition of the phlogopite. Take into consideration carbonates homogeneity and absence of their secondary phases we estimated crystallization temperature about 540 °C (according to the Talantsev's thermometer), and it is close to crystallization temperature of the coexisting carbonates according to isotopic calculations. Apatite of the deposit is rich in strontium and considerable variations of the strontium isotopes are characteristic. Isotopic composition of carbon and hydrogen from the Siillinjarvi carbonates are also variable, and often are not similar to the typical for young carbonatites.

*Key words:* carbonatite, igneous silicate rocks, apatite, tetraferriphlogopite, petrogenesis, isotopes of C, O, Sr, Siillinjarvi, Finland.

Carbonatites are relatively rare rocks composed mainly endogenic carbonates with some silicates and minor minerals of rare metals, REE, U and Th, etc. After first observations of carbonatites in the Fen complex, SE Norway [10], idea about their magmatic origin has been proposed. Since this recognition, a number of studies were performed on geology, mineralogy, origin and economic geology of carbonatites. Concentric and zonal shapes of the main carbonatite-bearing complexes are typical for their Phanerozoic varieties [3]. Paragenetic relationships of the ultrabasic rocks, alkaline rocks and carbonatites are indispensable conditions to establish real carbonatites in contrast to late carbonate veins, nests, and banded bodies, which are common inside zones of secondary alteration and hydrothermal halos. General similarity in mineral compositions and shapes carbonatites and late carbonates led to appearance a number of definitions to determine carbonatites, which combine different terms from magmatic up to metasomatic processes [1, 9, 12, 32]. Up to present time in the scientific literature we can meet harden discussions on the term “carbonatites”, and there are about ten versions of the term [15, 16, 19, 21, 25, 26]. In addition to primary structural, geological, and petrographic criterions to establish real carbonatites [3, 25] now geologists use geochemical, isotope and petrologic evidences for these purposes. For example, due to fundamental isotopic investigations of carbonates from carbonatites [14] limits of oxygen and carbon isotope marks have

been determined for typical carbonatites (“Oka box”). Besides clear evidence of carbonatite magmatic lavas from Oldoinyo Lengai, Tanzania [29] still now some geologists develop ideas on metasomatic origin of carbonatites, but another [23] note about success in investigation of relationship between carbonatite and kimberlite magmas. Different arguments on magmatic or metasomatic origin of carbonatites reflect real differences of mechanisms carbonatites formation from place to place in accordance to their evolution.

More complicated evidences we need to determine peculiarities of formation for the Precambrian carbonatites. Commonly they have linear shape, very often not all three rocks (ultrabasic, alkaline and carbonatites), which are typical for the Phanerozoic complexes, we can determine in association with carbonatites. Poorer specific for carbonatites rare metals mineralization, and wall rock alterations of the primary varieties are also problems for detection of the Precambrian carbonatites. Among the Precambrian carbonatites most old are the Siilliniarvi carbonatites, Eastern Finland [22, 28, 34, 40]. Since their discovery in 1950, they are examples of the Precambrian carbonatites due to old age and since 1979 – as a source of apatite and carbonate concentrates with some accompanies components for industry of Finland [28, 31, 37].

First idea on carbonatites origin evolved from transformation of limestone substratum and skarns [33, 35] up to modern points of view, which include mechanism of magma melting of the upper mantle [8, 29]. Basement of present models of carbonatite formation includes system of geological, mineralogical, and isotopic data [26, 27, 36, 38], and for the Precambrian carbonatites they are the only way to do conclusions on the carbonatites appearance.

To understand main peculiarities of the Siilliniarvi’s carbonatites origin we compared its characteristics with other possible Precambrian carbonatites of Siberia and of the Ukrainian Shield [4, 5, 7, 39].

**Geological setting and main rocks types.** Geological ideas on main rocks groups and structure of the Siiliniarvi deposit varied since the first systematic reports [31]. To describe carbonate rocks of the Siiliniarvi deposit as carbonatites, K. Puustinen [31] used the Heinrich’s definition [19] and pointed out for its purpose the association of ultramafic and alkaline rocks with carbonatites. This definition is close to opinion of A. Kukhareno and co-authors [3, 11] for the Phanerozoic carbonatites in close association with the ultrabasic and alkaline rocks, which are surrounding of fenites. But, in contrast to the Phanerozoic complexes with carbonatites, K. Puustinen [31] suggested glimmerites of the Siiliniarvi deposit as an analogous to the ultramafic rocks and indicated only sporadically development of fenites. In bigger scale at the scheme of distribution carbonatites and lamprophyres in the Fennoscandian Shield (Fig. 1) the Siiliniarvi carbonatites have been suggested as with no associated silicate rocks [37], and at the modern Yara’s map of the deposit we can see wider development of the fenites and absence of syenite (Fig. 2).

According to proposed schemes [31, 34] glimmerites of the deposit as primary ultramafic phases were intruded into the host granitic gneisses and system of the sub-vertical carbonate veins is developed in several stages inside the glimmerites. The carbonate veins jointly form main carbonatite tabular body about 16 km long and up to 1.5 km wide. It was assumed early [31] that intrusion of glimmerite was followed by intrusion of syenites in the marginal zone of the glimmerite and the surrounding bedrock, but later the syenitic rocks as well as glimmerites were regarded by P. Herms [34] as fenitized host rocks. The radiometric age of the complex ranges from 2 000 up to 2 900 million years, and the oldest formation age of carbonatites was inferred from U/Pb investigations on zircons as  $2.605 \pm 0.006$  Ga [20, 22, 31, 34, 40].

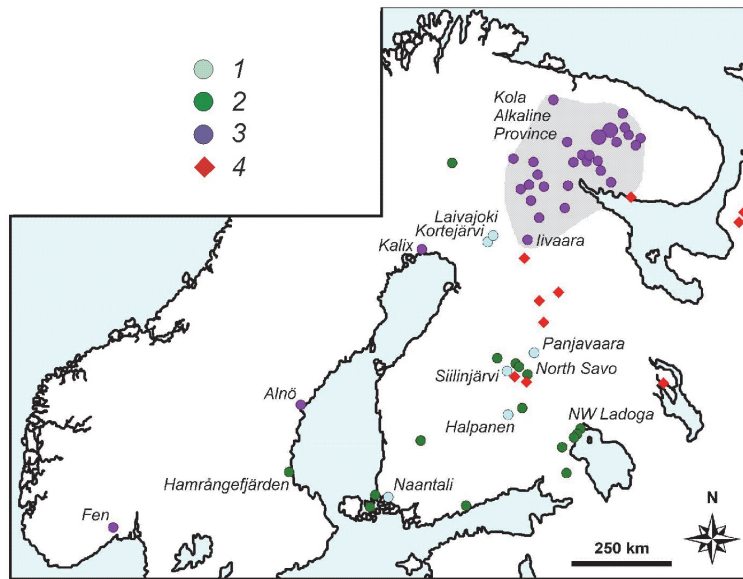


Fig. 1. The spatial distribution of carbonatites, lamprophyres and kimberlites in the Fennoscandian Shield [37]:

1 – carbonatites with no associated silicate rocks; 2 – shoshonitic lamprophyres with or without associated granitoids; 3 – ultramafic lamprophyres, carbonatites and associated alkaline silicate rocks; 4 – kimberlites of the first and second groups.

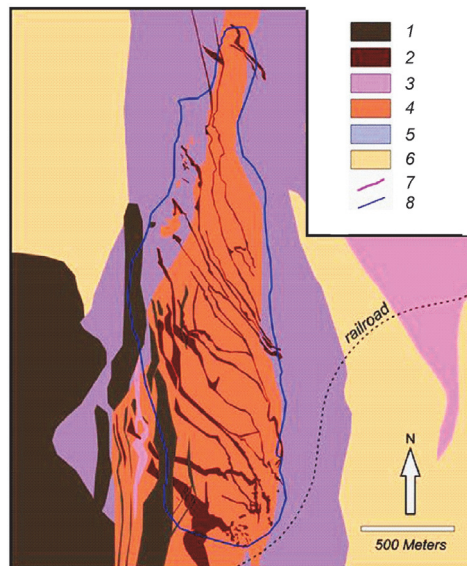


Fig. 2. Geological map of the Siilinjärvi carbonatite complex (the area of the main pit) [28].

1 – tonalite-diorite; 2 – diabase; 3 – granite; 4 – ore; 5 – fenite; 6 – gneiss; 7 – melasyenite dyke; 8 – current pit outline.

**Mineral composition of the rocks.** More than 25 minerals are established in the rocks of the Siillinjarvi deposit [28, 31, 37]. Besides dolomite and calcite (up to 70 % and more of total volume of the carbonatites), apatite, phlogopite, tetraferriphlogopite, richterite are common components of the rocks.

Sometimes white carbonatites (Fig. 3) contain rich in colourless apatite and silicates strips and nests more dark in colour. Phlogopite, richterite are main minerals of the glimmerites, and carbonates (less than 50 %) and apatite and zircon are minor phases [31].



Fig. 3. Contact of light carbonatite and carbonatite rich in tetraferriphlogopite.

In contrast to small grains of apatite in carbonatites, in parts of dark and rich in tetraferriphlogopite glimmerites apatite grains are bigger in size and gray-green in color (Fig. 4). Strontianite, monazite, pyrochlore, baddeleyite, ilmenite, magnetite and pyrite are present as accessory minerals in the rocks [31]. Microcline with small amount of albite is minor minerals in the syenites, and absence of nepheline. Dark minerals are represented by aegirine-augite and actinolite. Some pegmatoid parts of syenites contain titanite rich in Nb. There is also information about finds of fergusonite and columbite [31].

Siillinjarvi carbonatite complex is one of the oldest carbonatite deposits in the world with in situ grade 4.2 %  $P_2O_5$  with automated XRF/XRD for on-line analyses in the concentrator. Siillinjarvi Phosphate Mine currently owned and operated by Yara. Production of the apatite concentrates started in 1979 and has been in continuous operation since then firstly from the Sarkijarvi pit and from the Saarinen pit later (2012).

**Carbonates.** According to observations with thin sections dolomite and calcite varieties of the carbonatites of the Siillinjarvi deposit are developed independently. We did not found any substitutions between calcites and dolomites in the carbonatites, and we assumed that chemical compositions of the carbonates reflect its primary phases.

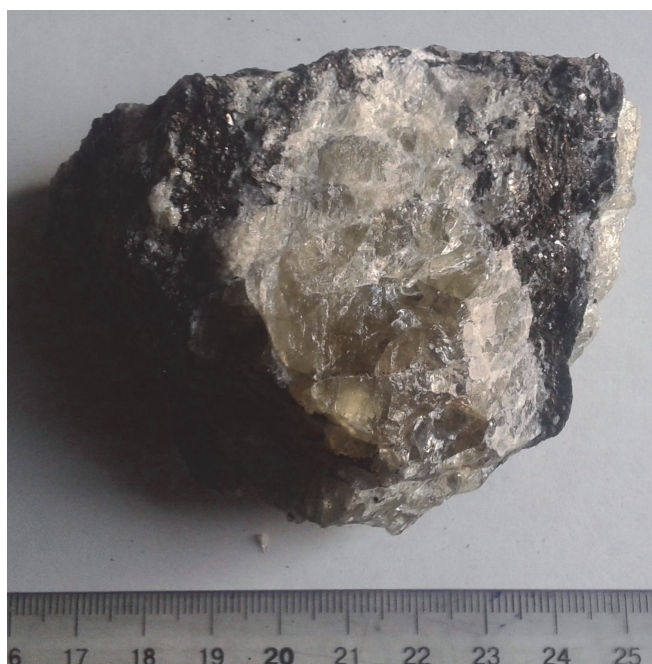


Fig. 4. Apatite crystal surrounding by tetraferriphlogopite rose.

Chemical compositions of the carbonates (Table 1) are relatively homogenous, and at the compositional diagram (Fig. 5) we can see two isolated fields – one of the pure calcites, and second – of dolomites. Taking into consideration homogeneity of the carbonates and absence of secondary phases, we calculated its crystallization temperature with the Talantsev's geothermometer [6] approximately under 540 °C. It is close to the crystallization temperatures obtained by calculations after isotopic data for the coexisting carbonates [13].

**Tetraferriphlogopite.** During investigations of the tetraferriphlogopite of the Siillinjarvi deposit in thin sections, we found its high homogeneity. Its chemical compositions (Table 2) are relatively stable and at the compositional diagram (Fig. 6), the tetraferriphlogopites form compact and isolated field. Obtained data confirm previous conclusions [30, 31] about absence of alteration of the primary phlogopite.

**Apatite.** In the Siillinjarvi deposit the mineral occurs mainly as irregular grains up to a few millimetres in carbonatites and glimmerites. Sometimes apatite grains up to a few centimetres with vague crystallographic forms are present in glimmerites surrounding by aggregates of tetraferriphlogopite (See Fig. 2).

Main peculiarities of chemical composition of apatite from the Siillinjarvi deposit are deficiency of phosphorus and high strontium concentration (Table 3) [31]. Last is typical for apatite from the Phanerozoic alkaline rocks (up to 5.5 % SrO in Murun and Khibiny massifs) and carbonatites.

At the binary diagram (Fig. 7), which reflects strontium and calcium isomorphic substitutions, high strontium concentrations in apatite from the Precambrian linear carbonatites of Ukraine [5, 39] and apatite poor in strontium from the Precambrian carbonate-bearing rocks [17] are shown.

Table 1  
Chemical composition and formula coefficients of carbonates from Siilinjarvi deposit

| Com-<br>po-<br>nents | Number of the analysis |       |       |       |       |       |       |       |       |       |       |
|----------------------|------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
|                      | 1                      | 2     | 3     | 4     | 5     | 6     | 7     | 8     | 9     | 10    | 11    |
| MgO                  | 0.52                   | 0.38  | 0.59  | 17.48 | 17.46 | 17.35 | 17.72 | 17.34 | 17.82 | 1.55  | 1.66  |
| CaO                  | 55.02                  | 56.84 | 54.26 | 31.15 | 30.34 | 32.40 | 30.63 | 29.87 | 32.66 | 52.84 | 51.30 |
| MnO                  | 0.19                   | 0.65  | 0.29  | 0.74  | 0.12  | 0.51  | 0.30  | 0.39  | 0.37  | 0     | 0.63  |
| FeO                  | 0.76                   | 0.59  | 0.68  | 3.85  | 3.02  | 3.36  | 3.17  | 2.83  | 3.25  | 0.63  | 0.25  |
| NiO                  | 0.49                   | 0     | 0.64  | 0     | 0     | 0     | 0     | 0.20  | 0     | 0.46  | 0.39  |
| SrO                  | 1.52                   | 0.63  | 0.75  | 0.74  | 0     | 1.34  | 0.50  | 0.52  | 0.73  | 1.64  | 1.17  |
| Σ                    | 58.49                  | 59.09 | 57.21 | 53.96 | 50.94 | 54.96 | 52.32 | 51.15 | 54.83 | 57.12 | 55.40 |
| Ca                   | 0.95                   | 0.97  | 0.96  | 1.05  | 1.06  | 1.07  | 1.05  | 1.05  | 1.08  | 0.93  | 0.93  |
| Mg                   | 0.01                   | 0.01  | 0.01  | 0.82  | 0.85  | 0.80  | 0.85  | 0.85  | 0.82  | 0.04  | 0.04  |
| Mn                   | 0.00                   | 0.01  | 0.00  | 0.02  | 0.00  | 0.01  | 0.01  | 0.01  | 0.01  | 0.00  | 0.01  |
| Fe <sup>2+</sup>     | 0.01                   | 0.01  | 0.01  | 0.10  | 0.08  | 0.09  | 0.08  | 0.08  | 0.08  | 0.01  | 0.00  |
| Sr                   | 0.01                   | 0.01  | 0.01  | 0.01  | 0.00  | 0.02  | 0.01  | 0.01  | 0.01  | 0.02  | 0.01  |
| Ni                   | 0.01                   | 0.00  | 0.01  | 0.00  | 0.00  | 0.00  | 0.00  | 0.01  | 0.00  | 0.01  | 0.01  |
| Σ                    | 1                      | 1     | 1     | 2     | 2     | 2     | 2     | 2     | 2     | 1     | 1     |

Table 2  
Chemical composition and formula coefficients of tetraferroflogopite from Siilinjarvi deposit

| Components                     | Number of the analysis |       |       |       |       |
|--------------------------------|------------------------|-------|-------|-------|-------|
|                                | 1                      | 2     | 3     | 4     | 5     |
| SiO <sub>2</sub>               | 40.22                  | 39.98 | 40.44 | 40.54 | 40.54 |
| Al <sub>2</sub> O <sub>3</sub> | 6.69                   | 6.83  | 6.26  | 6.73  | 6.88  |
| TiO <sub>2</sub>               | 0.22                   | 0.42  | 0.17  | 0.39  | 0.34  |
| FeO                            | 6.80                   | 9.00  | 7.00  | 6.80  | 7.20  |
| Fe <sub>2</sub> O <sub>3</sub> | 6.88                   | 4.85  | 6.79  | 7.56  | 6.54  |
| MgO                            | 22.61                  | 21.85 | 22.78 | 23.23 | 23.06 |
| MnO                            | 0.20                   | 0.36  | 0     | 0.37  | 0.40  |
| CaO                            | 0.03                   | 0.08  | 0.01  | 0.14  | 0     |
| Na <sub>2</sub> O              | 0                      | 0     | 0     | 0     | 0.15  |
| K <sub>2</sub> O               | 10.93                  | 10.96 | 10.74 | 10.87 | 11.32 |
| Σ                              | 94.58                  | 94.33 | 94.19 | 96.63 | 96.43 |
| Si                             | 3.01                   | 3.23  | 3.04  | 2.98  | 2.99  |
| Al                             | 0.59                   | 0.65  | 0.55  | 0.58  | 0.60  |
| Ti <sup>4+</sup>               | 0.01                   | 0.03  | 0.01  | 0.02  | 0.02  |
| Fe <sup>3+</sup>               | 0.39                   | 0.29  | 0.38  | 0.42  | 0.36  |
| Σ                              | 4.00                   | 4.19  | 3.98  | 4.00  | 3.97  |
| Fe <sup>2+</sup>               | 0.43                   | 0.61  | 0.44  | 0.42  | 0.44  |
| Mg                             | 2.52                   | 2.63  | 2.55  | 2.54  | 2.53  |
| Mn <sup>2+</sup>               | 0.01                   | 0.02  | 0.00  | 0.02  | 0.02  |
| Σ                              | 2.96                   | 3.26  | 2.99  | 2.98  | 3.00  |
| Ca                             | 0.00                   | 0.01  | 0.00  | 0.01  | 0.00  |
| Na                             | 0.00                   | 0.00  | 0.00  | 0.00  | 0.02  |
| K                              | 1.04                   | 1.13  | 1.03  | 1.02  | 1.06  |
| Σ                              | 1.05                   | 1.13  | 1.03  | 1.03  | 1.09  |

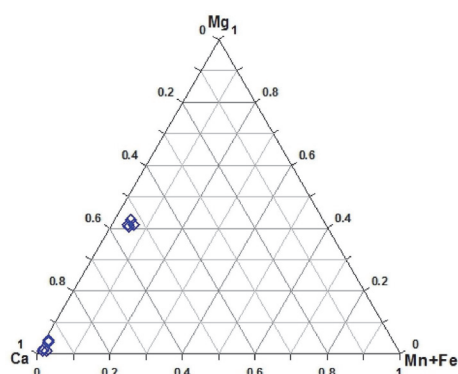


Fig. 5 Variations of the Siilinjärvi's carbonates compositions on the diagram Ca–Mg–(Mn<sup>2+</sup>+Fe<sup>2+</sup>).

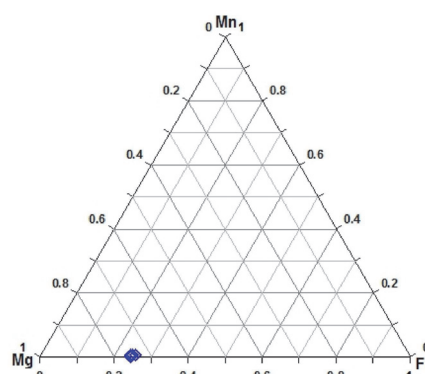


Fig. 6. Variations of the Siilinjärvi's tetraferryflogopites compositions on the diagram Mg–Mn–Fe.

Table 3

Chemical composition and formula coefficients of apatite from Siilinjärvi deposit

| Components                    | Number of the analysis |       |       |       |
|-------------------------------|------------------------|-------|-------|-------|
|                               | 1                      | 2     | 3     | 4     |
| P <sub>2</sub> O <sub>5</sub> | 38.11                  | 37.06 | 37.56 | 38.31 |
| CaO                           | 56.40                  | 55.23 | 55.12 | 56.22 |
| K <sub>2</sub> O              | 0                      | 0.37  | 0.10  | 0.11  |
| MnO                           | 0.27                   | 0     | 0     | 0     |
| MgO                           | 0.22                   | 0     | 0     | 0     |
| FeO                           | 0.13                   | 0     | 0     | 0     |
| Σ                             | 95.14                  | 92.66 | 92.79 | 94.64 |
| P <sub>2</sub> O <sub>5</sub> | 2.96                   | 2.96  | 2.98  | 2.98  |
| Ca                            | 5.54                   | 5.58  | 5.54  | 5.54  |
| Mn                            | 0.02                   | 0.00  | 0.00  | 0.00  |
| Mg                            | 0.03                   | 0.00  | 0.00  | 0.00  |
| Fe                            | 0.01                   | 0.00  | 0.00  | 0.00  |
| K                             | 0.00                   | 0.04  | 0.01  | 0.01  |
| Σ                             | 5.60                   | 5.63  | 5.55  | 5.55  |

Isotopic compositions of Sr in apatite of the Siilinjärvi deposit are variable [13, 34] and are close to the crust marks in apatite from the glimmerites and the Precambrian metasedimentary carbonate rocks [2, 17, 24]. At the same time the lowest Sr isotope ratios of apatites and carbonates from the Siilinjärvi carbonatites (0.70137) are in the range of bulk silicate Earth and higher than those from Canadian carbonatites of similar ages [20, 34].

System investigations of the geology, mineralogy and petrology of the Siilinjärvi deposit [20, 28, 31, 40] has been shown, on one hand, the presence of carbonatites in the Fennoscandian Shield Archean rocks, which are characteristic for more young, Phanerozoic complexes, and on the other hand – their diversity by linear form, reduced and specific in composition ultramafic and alkaline rocks, common for ultrabasic-alkaline complexes with carbonatites, poor in minerals with rare metals and REE specialization or its absence, etc.

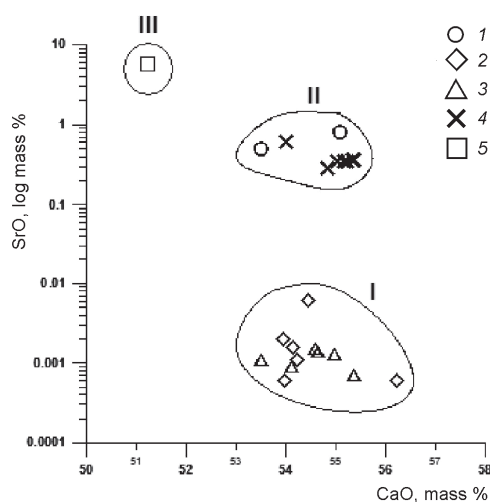


Fig. 7. Variations of apatite compositions on the diagram CaO–SrO:

1 – from calcite carbonatites of the Behim-Chokrak (Chernihiv zone, Pre-Azov region, Ukraine); 2 – from the Precambrian Mg-skarn; 3 – from the Precambrian apatite bearing carbonate rocks; 4 – from kimberlites; 5 – from alkaline metasomatic rocks of the Murun massif (North-Western Aldan, Siberia). Fields of the apatite of similar chemical composition: I – from the Precambrian Mg-skarns and apatite bearing carbonate rocks; II – from kimberlites and calcite carbonatites of the Behim-Chokrak; III – from alkaline metasomatic rocks of the Murun massif.

Exploration and mining for some decades of the Siillinjarvi deposit in open pits promoted in news data about the deposit [28, 37], but final decision on origin of the carbonatites and associated rocks are still far.

Modern data on peculiarities of geological and structural position as well as chemical and mineralogical compositions of the rocks and minerals of the Siillinjarvi deposit can be used as special etalon for the Precambrian linear carbonatites to estimate evolution of Precambrian endogenous formations. There is similarity of the Siillinjarvi deposit and other Precambrian deposits [5, 7, 39] by linear shape of the main carbonatite bodies, but there is not direct analogous in composition and transformation of the primary rocks of the deposits. Glimmerites are regarded as products of alteration of ultramafic initial rocks, but there is a big homogeneity among tetraferriphlogopite grains. The silicate rocks of the Siillinjarvi deposit have close chemical compositions to typical relatively fresh ultramafic rocks, but we cannot retrace all line of transformations in the glimmerites. Similar peculiarity is characteristic of the carbonates.

Minerals of rare metals and REE, characteristic of the Phanerozoic carbonatites, have been found only in small amounts or are absent at all. Besides high strontium concentration in apatite another strontium minerals typical for carbonatites are very rare in the Siillinjarvi deposit.

Isotopic compositions of carbon and oxygen in carbonates of the Siillinjarvi deposit are also heterogeneous and deflected from typical for limits of young carbonatite [14]. Relatively high data for isotopically heavy carbon ( $\delta^{13}\text{C} = -3.7\text{‰}$ , PDB) from the carbonatites are close to characteristics which we found in carbonates formed in near surface conditions [18].



REFERENCES

1. Багдасаров Ю. А. О принципиальной возможности определения критериев прогнозирования и оценки рудоносности карбонатитовых массивов / Ю. А. Багдасаров // Разведка и охрана недр. – 2007. – № 4. – С. 16–22.
2. Генетические особенности апатит-карбонатных пород селигдарского типа (Алдан) в свете новых данных по вариациям изотопов углерода и стронция / А. Р. Энтин, А. И. Зайцев, Н. И. Ненашев [и др.] // Докл. АН СССР. – 1987. – Т. 287, № 5. – С. 1228–1232.
3. Каледонский комплекс ультраосновных, щелочных пород и карбонатитов Кольского полуострова и Северной Карелии / [А. А. Кухаренко, М. П. Орлова, А. Г. Булах и др.]. – М. : Недра, 1965. – 772 с.
4. Кривдик С. Г. О связи карбонатитового и кимберлитового магматизма (на примере докембрийского карбонатитового комплекса Приазовья / С. Г. Кривдик // Геохимия и рудообразование. – 1980. – № 8. – С. 66–78.
5. Кривдик С. Г. Щелочной магматизм Приазовья / С. Г. Кривдик, Н. В. Безсмолова, А. В. Дубина // Наук. праці УкрНДМІ НАН України. – 2009. – № 5, ч. 2. – С. 158–166.
6. Таланцев А. С. Геотермобарометрия по доломит-кальцитовым парагенезисам / А. С. Таланцев. – М. : Наука, 1989. – 200 с.
7. Alkaline Rocks and Carbonatites of the World. Part 2: Former USSR / L. N. Kogarko, V. A. Kononova, M. P. Orlova, A. R. Woolley. – London ; New York : Chapman and Hall, 1995. – 226 p.
8. Blichert-Toft J. Precambrian alkaline magmatism / J. Blichert-Toft, N. Arndt, J. Ludden // Lithos. – 1996. – Vol. 37. – P. 97–111.
9. Brassinnes S. Magmatic evolution of the differentiated ultramafic, alkaline and carbonatite intrusion of Vuoriyarvi (Kola Peninsula, Russia). A LA-ICP-MS study of apatite / S. Brassinnes, E. Balaganskaya, D. Demaiffe // Lithos. – 2005. – Vol. 85. – P. 76–92.
10. Brogger W. C. Die Eruptivgesteine des Kristianiagebietes, IV. Das Fengebiet in Telemark, Norwegen / W. C. Brogger. – Kristiania, 1921. – 408 p.
11. Bulakh A. G. Overview of carbonatite-phoscorite complexes of the Kola Alkaline Province in the context of a Scandinavian North Atlantic Alkaline Province / A. G. Bulakh, V. V. Ivanikov, M. P. Orlova // Phoscorites and Carbonatites from Mantle to Mine: The Key Example of the Kola Alkaline Province. – The Mineralogical Society of Great Britain & Ireland, 2004. – P. 1–43.
12. Burke K. Geoinformatic approach to global nepheline syenite and carbonatite distribution: Testing a Wilson cycle model / K. Burke, S. Khan // Geosphere. – 2006. – Vol. 2, N 1. – P. 53–60.
13. Carbonatite metasomatism: evidence from geochemistry and isotope composition (U-Pb, Hf, O) on zircons from two Precambrian carbonatites of the Kola alkaline province / M. Tichomirowa, M. Whithehouse, A. Gerdeos, J. Götze // Goldschmidt Conference, 2012. – P. 875.
14. Deines P. Stable isotope variations in carbonatites / P. Deines // Carbonatites – Genesis and Evolution. – London : Unwin Hyman, 1989. – P. 301–359.

15. Fourcade S. Metasomatic effects related to channelled fluid streaming through deep crust: fenites and associated carbonatites (In Ouzzal Proterozoic granulites, Hoggar, Algeria) / S. Fourcade, J.-R. Kienast, K. Oozegane // *Journal of Metamorphic Geology*. – 1996. – Vol. 14, N 6. – P. 763–781.
16. Green D. H. Mantle metasomatism by ephemeral carbonatite melts / D. H. Green, M. E. Wallace // *Nature*. – 1988. – Vol. 336. – P. 459–462.
17. Guliy V. N. Main features of composition and origin of apatite deposits in metamorphic rocks of the Aldan Shield, Siberia, Russian Federation / V. N. Guliy // *Trans. Inst. Min. Metall.* – 1995. – Vol. 104. – P. B171–B178.
18. Guliy V. N. Carbon and oxygen isotopic compositions of carbonates from Precambrian apatite-bearing carbonate rocks of the Aldan Shield / V. N. Guliy, H. Wada // *Lithology and Mineral Resources*. – 2004. – Vol. 39. – P. 243–253.
19. Heinrich E. W. *The Geology of Carbonatites* / E. W. Heinrich. – Chicago : Rand McNally & Co., 1966. – 555 p.
20. Isotopic compositions of noble gas and carbon in the Archean carbonatites from the Siilinjärvi mine, central Finland / Matsumoto Takuya, Maruoka Teruyuki, Matsuda Junichi [et al.] // *Goldschmidt Conference, 2008*. – P. 873.
21. Kapustin Yu. L. On the origin of carbonatites / Yu. L. Kapustin // *Intern. Geol. Rev.* – 1977. – Vol. 19. – P. 997–1008.
22. Karhu J. A. Radiometric ages and isotope systematics of some Finnish carbonatites / J. A. Karhu, I. Manttari, H. Huhma // *Res. Terrae*. – 2001. – Ser. A. – Vol. 19. – P. 8.
23. Kjarsgaard B. A. The genesis of carbonatites by immiscibility / B. A. Kjarsgaard, D. L. Hamilton // *Carbonatites – Genesis and Evolution*. – London : Unwin Hyman, 1989. – P. 388–404.
24. Lead- and strontium-isotope composition of apatite-silicate and apatite-carbonate ores in Aldan Shield / Yu. D. Pushkarev, V. N. Guliy, M. P. Kravchenko, G. I. Ryungenen // *12 Всесоюз. симпозиум по стабильным изотопам в геохимии : тез. докл.* – М., 1989. – С. 10–11.
25. Le Bas M. J. Diversification of carbonatite / M. J. Le Bas // *Carbonatites – Genesis and Evolution*. – London : Unwin Hyman, 1989. – P. 428–447.
26. Mitchell R. M. Carbonatites and carbonatites and carbonatites / R. M. Mitchell // *Can. Mineral.* – 2005. – Vol. 43, N 6. – P. 2049–2068.
27. New evidence from a calcite-dolomite carbonatite dyke for the magmatic origin of the massive Bayan Obo ore-bearing dolomite marble, Inner Mongolia, China / M. J. Le Bas, Xueming Y., R. N. Taylor [et al.] // *Mineralogy and Petrology*. – 2007. – Vol. 90. – P. 223–248.
28. O'Brien H. The Archean Siilinjärvi carbonatite complex / H. O'Brien, E. Heilimo, P. Heino // *Mineral Deposits of Finland*. – Amsterdam : Elsevier, 2015. – P. 327–343.
29. Origin of carbonatites by liquid immiscibility / D. L. Hamilton, I. C. Freestone, J. B. Dawson, C. H. Donaldson // *Nature*. – 1979. – Vol. 279. – P. 52–54.
30. Poutiainen M. Fluids in the Siilinjärvi carbonatite complex, eastern Finland: Fluid inclusion evidence for the formation conditions of zircon and apatite / M. Poutiainen // *Bull. Geol. Soc. Finland*. – 1995. – Vol. 67, part 1. – P. 3–18.
31. Puustinen K. Geology of the Siilinjärvi carbonatite complex, Eastern Finland / K. Puustinen // *Bulletin de la Commission Geologique de Finlande*. – 1971. – N 249. – 43 p.

32. Subducted carbonates, metasomatism of mantle wedges, and possible connections to diamond formation: An example from California / M. N. Ducea, J. Saleeby, J. Morrison, V. Valencia // *Amer. Mineral.* – 2005. – Vol. 90, N 5–6. – P. 864–870.
33. Suomen kalkkikivi. Summary: Limestones in Finland / P. Eskola, V. Hackman, A. Laitakari, W. W. Wilkman // *Geologinen tutkimuslaitos, Geoteknillisi julkaisuja.* – 1919. – Vol. 21. – P. 127–128.
34. The mineral isotope composition of two Precambrian carbonatite complexes from the Kola alkaline province – Alteration versus primary magmatic signatures / M. Tichomirowa, G. Grosche, J. Götze [et al.] // *Lithos.* – 2006. – Vol. 91. – P. 229–249.
35. Watkinson D. H. Phase equilibrium studies bearing on the limestone-assimilation hypothesis / D. H. Watkinson, P. J. Wyllie // *Geol. Soc. of Amer. Bull.* – 1969. – Vol. 80. – P. 1563–1576.
36. Wolley A. R. The spatial and temporal distribution of carbonatites / A. R. Wolley // *Carbonatites – Genesis and Evolution.* – London : Unwin Hyman, 1989. – P. 15–37.
37. Woodard J. Genesis and Emplacement of Carbonatites and Lamprophyres in the Svecofennian Domain : Academic Dissertation / J. Woodard. – Turku : Turun Yliopisto University of Turku, 2010. – 50 p.
38. Woolley A. R. Igneous silicate rocks associated with carbonatites: their diversity, relative abundances and implications for carbonatite genesis / A. R. Woolley // *Per. Mineral.* – 2003. – Vol. 72. – P. 9–17.
39. Zagnitko V. M. Isotopic geochemistry of carbonatites of Ukraine / V. M. Zagnitko, S. G. Kryvdik, M. O. Donskiy // *Per. Mineral.* – 2003. – Vol. 72. – P. 153–159.
40. Age and genesis of the Siilinjarvi Archean carbonatite complex in light of isotope data / B. V. Belyatsky, E. V. Savva, M. Tikhomirova [et al.] // 11<sup>th</sup> Annual V. M. Goldschmidt Conf., 2001. [Electronic resource. Available at: 3622.pdf].

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## **ОСОБЛИВОСТІ МІНЕРАЛЬНОГО СКЛАДУ ТА ПОХОДЖЕННЯ КАРБОНАТИТІВ СІЛІНЬЯРВІ (ФІНЛЯНДІЯ)**

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Висвітлено результати дослідження докембрійських карбонатитів (альвікіти і бефорсити) родовища Сіліньярві (Фінляндія). Карбонатити вкорінені в давніші флогопітові породи – сієніти і глімерити. Головними мінералами є доломіт і кальцит (становлять понад 70 % від загального об'єму карбонатитів); серед другорядних виявлено апатит, тетраферифлогопіт і рихтерит, серед акцесорних – стронціаніт, монацит, пірохлор, баделеїт, ільменіт, магнетит, пірит. Визначено однорідний хімічний склад флогопіту. На підставі однорідності карбонатів і відсутності їхніх вторинних фаз з'ясовано температуру їхньої кристалізації (за

геотермометром А. Таланцева) – 540 °С, що близько до температури кристалізації карбонатів, що співіснують, за ізотопними значеннями. Досліджений апатит збагачений стронцієм за значних варіацій співвідношення ізотопів стронцію. Ізотопний склад вуглецю й кисню в карбонатах родовища Сіліньярві також неоднорідний і часто не відповідає типовому для карбонатів із карбонатитів.

*Ключові слова:* карбонатити, магматичні силікатні породи, апатит, тетраферифлопогіт, петрогенезис, ізотопи С, О, Sr, Сіліньярві, Фінляндія.